Note

Regioselectivity in the reactivity of dibutylstannylene derivatives of glycals*

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(Received March 2nd, 1988; accepted for publication, October 7th, 1988)

Regioselective acylations of D-glucal (1) have been studied intensively¹ due to their utility in synthesis. Conventional methods of acylation of 1 yield all possible esters¹. Hindered reagents (TsCl², Ph₃CCl³, 'BuMe₂SiCl⁴⁻⁶) selectively acylate at O-6. Regioselective acylations⁷ and silylations⁶ of L-rhamnal (14) gave 3- or 4-acyl (40–60%) and 3-monosilylated (70–95%) derivatives, respectively. Cyclic tin derivatives of vicinal diols show selectivity in various reactions⁸. Thus, regioselective acetylations⁸⁻¹⁰, alkylations^{11–13}, and oxidations^{14–16} of stannylene derivatives of carbohydrates and the monosubstitution of 2',3'-O-(dibutylstannylene)-nucleosides⁸ have been reported.

In a search for improved selectivity, the acylation of cyclic dibutylstannylene derivatives of some glycals has been investigated. It was anticipated that the flexible dihydropyran ring would permit the formation of a cyclic dibutylstannylene derivative, and we now report a practical method for 3-acylation of the dibutylstannylene derivatives 7 and 15.

Reaction of **1** with 1.1 mol. equiv. of dibutyltin oxide in hot methanol resulted in dissolution, usually within 40 min. Since acylations using this solution did not go to completion even after the addition of a large excess of the acylating agents⁸, the methanol was removed and the resulting white solid (**3/4**) was acylated. The dibutylstannylene derivatives (**7** and **15**, respectively) of 6-*O*-trityl-D-glucal³ (**2**) and L-rhamnal¹⁷ (**14**) were prepared similarly. The 3-acylated products (**6**, **8**, **11**, **16**, and **19**) obtained in these reactions were purified by column chromatography and their structures established on the basis of the ¹H- and ¹³C-n.m.r. spectra and by comparison with the data of known compounds^{6,7}.

The reaction of 3/4 with Ac_2O/NEt_3 (1 mol. equiv. of each), either in N,N-dimethylformamide or dichloromethane for 6 h at room temperature, gave 53% of a 7:2 mixture (n.m.r. data) of 6- (5) and 3-O-acetyl-D-glucal (6) which could not be

^{*}NCL Communication No. 4399.

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fractionated. Similar mixtures were obtained on reaction of 3/4 with AcCl or BzCl/ NEt₃. The reaction of 1 with Ac₂O/pyridine for 6 h at room temperature yielded a 2.5:1 mixture of 5 and 6. Thus, it appears that a mixture of 4,6- (3) and 3,4-di-O-(dibutylstannylene)-D-glucal (4) is probably involved in these reactions, so that selective acylation of 1 was not possible and, therefore, 6-O-trityl-D-glucal (2) was investigated.

Reaction of the 2,3-O-dibutylstannylene derivative (7) of 2 with Ac₂O or BzCl/NEt₃ (1.5 mol. equiv.) in N,N-dimethylformamide or dichloromethane for 3 h at room temperature gave mainly the 3-O-acetyl (8, 88%) and 3-O-benzoyl (11, 85%) derivatives, respectively. Likewise, treatment of the L-rhamnal derivative 15 with Ac₂O or BzCl/NEt₃ for 4 h gave the 3-O-acetyl⁷ (16, 81%) and 3-O-benzoyl⁷ (19, 78%) derivatives, respectively. The absence of 4-O-acyl derivatives (9, 177, and 207) in these reactions was confirmed by t.l.c., using reference compounds prepared by non-selective methods. Partial acylation⁷ of 2 gave 8-11 and 13 (12 was not formed). Compounds 17 and 20 were obtained in a similar manner⁷. Acylation at O-3 and O-4 resulted in a downfield shift (1.0-1.2 p.p.m.) of the signals for H-3 (ddd) and H-4 (dd), respectively, in the ¹H-n.m.r. spectra. Likewise, there were shifts⁷ in the resonances for C-1 (downfield by 1.3-2.0 p.p.m.) and C-2 (upfield by 4.0-5.0 p.p.m.) in the ¹³C-n.m.r. spectra on acylation at O-3. Only small shifts occurred on 4-acylation. The dibutylstannylene derivatives 3/4, 7, and 15 did not react under alkylating conditions (MeI or PhCH₂Br in N, N-dimethylformamide at 30-80°), and higher temperature and longer reaction time caused decomposition of the enol ethers.



1
$$R^1 = R^2 = R^3 = H$$

2 $R^1 = R^2 = H, R^3 = Tr$
3 $R^1 = H, R^2, R^3 = SnBu_2$
4 $R^1, R^2 = SnBu_2, R^3 = H$
5 $R^1 = R^2 = H, R^3 = Ac$
6 $R^1 = Ac, R^2 = R^3 = H$
7 $R^1, R^2 = SnBu_2, R^3 = Tr$
8 $R^1 = Ac, R^2 = H, R^3 = Tr$
9 $R^1 = H, R^2 = Ac, R^3 = Tr$
10 $R^1 = R^2 = Ac, R^3 = Tr$
11 $R^1 = Bz, R^2 = H, R^3 = Tr$

12 $R^1 = H_1R^2 = Bz_1R^3 = Tr$ 13 $R^1 = R^2 = Bz_1R^3 = Tr$



15
$$R^4$$
, $R^5 = SnBu_2$
16 $R^4 = Ac$, $R^5 = H$
17 $R^4 = H$, $R^5 = Ac$
18 $R^4 = R^5 = Ac$
19 $R^4 = Bz$, $R^5 = H$
20 $R^4 = H$, $R^5 = Bz$
21 $R^4 = R^5 = Bz$

14 $R^4 = R^5 = H$

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EXPERIMENTAL

General methods. — N.m.r. spectra (¹H, 90 MHz; ¹³C, 22.63 MHz) were recorded for solutions in CDCl₃ (internal Mc₄Si) with a Bruker WH-90-FT spectrometer. Optical rotations were measured on a JASCO-DIP-181 polarimeter. Silica Gel 60 (230–400 mesh ASTM) (Merck) was used for flash-column chromatography. T.l.c. was performed on Silica Gel G (Acme). Compounds were purified by column chromatography.

Preparation of the dibutylstannylene derivatives. — To a solution of 1, 2, or 14 (5 mmol) in methanol (50 mL) was added dibutyltin oxide (5.5 mmol), and the solution was heated to reflux until it became homogeneous (40 min). The solvent was evaporated and the resulting dibutylstannylene derivative (3/4, 7, or 15) was dried under high vacuum (60° for 1 h) before use.

- 3- (6) and 6-O-acetyl-D-glucal (5). A suspension of 3/4 (0.3 g, 0.8 mmol) in dry N, N-dimethylformamide was treated with Et₃N (0.15 mL, 1 mmol) and Ac₂O (0.1 mL, 1 mmol) for 6 h at room temperature. Solvent was removed from the resulting solution under high vacuum. Column chromatography of the residue on silica gel (60–120 mesh, 10 g), using light petroleum–ethyl acetate (9:1), gave a 7:2 mixture (78 mg, 52%) of 5 and 6. ¹H-N.m.r. data: δ 2.08, 2.15 (2 s, 6 H, 2 OAc), 2.30–2.95 (m, 3 H, HO-3,4,6), 3.44–4.10 (m, 8 H, H-4,5,6,6'), 4.28, 5.29 (ddd, 2 H, $J_{2,3}$ 2, $J_{3,4}$ 7, $J_{1,3}$ 1.7 Hz, H-3), 4.67, 4.75 (dd, 2 H, $J_{1,2}$ 6, $J_{2,3}$ 2 Hz, H-2), 6.32, 6.44 (dd, 2 H, H-1).
- 3- (8) and 4-O-acetyl- (9), 3,4-di-O-acetyl- (10), 3-O-benzoyl- (11), and 3,4-di-O-benzoyl-6-O-trityl-D-glucal (13). To a solution of 2³ (1.18 g, 3 mmol) in N,N-dimethylformamide (8 mL) and pyridine (1 mL, 12.5 mmol) at 0° was slowly added Ac₂O (0.32 mL, 3 mmol) or BzCl (0.35 mL, 3 mmol). The solution was then stirred for 4 h at room temperature, diluted with water, and extracted with dichloromethane, the combined extracts were concentrated, and the residue was subjected to flash chromatography (light petroleum-ethyl acetate, 3:1) to yield 10 (282 mg, 19.7%), 8 (143 mg, 11%), 9 (286 mg, 22%), 13 (700 mg, 39%), and 11 (675 mg, 45%).

A suspension of 7 (1.24 g, 2 mmol) in N,N-dimethylformamide (3 mL; or CH_2Cl_2 , 5 mL) was treated with Et_3N (0.45 mL, 3.2 mmol) and Ac_2O (0.3 mL, 3 mmol) or BzCl (0.35 mL, 3 mmol) for 2 h at room temperature. Solvent was removed and the residue was subjected to column chromatography (light petroleumethyl acetate, 20:1) to give **8** (0.84 g, 85%) and **11** (0.76 g, 88%) as syrups.

Compound **2** had $[\alpha]_D$ +20° (c 1, methanol). N.m.r. data: 1 H, δ 2.15 (bs, 2 H, HO-3,4), 3.31 (dd, 1 H, $J_{6,6'}$ 10, $J_{5,6}$ 4 Hz, H-6), 3.51 (dd, 1 H, $J_{5,6'}$ 2.5 Hz, H-6'), 3.70–3.93 (m, 2 H, H-4,5), 4.11–4.33 (m, 1 H, H-3), 4.73 (dd, 1 H, $J_{1,2}$ 6, $J_{2,3}$ 2 Hz, H-2), 6.40 (dd, 1 H, H-1), 7.10–7.66 (m, 15 H, 3 Ph); 13 C, δ 63.2 (t, C-6), 69.8, 71.3, 77.1 (3 d, C-3,4,5), 87.2 (s, CPh₃), 102.8 (d, C-2), 127.4, 128.1, 128.9 (aromatic), 143.9 (s, aromatic), 144.6 (d, C-1).

Compound 8 had $[\alpha]_D$ +2° (c 0.26, methanol). N.m.r. data: ¹H, δ 2.08 (s, 3)

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H, OAc), 2.84 (d, 1 H, J 5 Hz, HO-4), 3.28–3.66 (m, 2 H, H-6,6'), 3.77–4.22 (m, 2 H, H-4,5), 4.73 (dd, 1 H, $J_{1,2}$ 6, $J_{2,3}$ 2 Hz, H-2), 5.26 (ddd, 1 H, $J_{1,3}$ 1.7, $J_{3,4}$ 8 Hz, H-3), 6.51 (dd, 1 H, H-1), 7.11–7.66 (m, 15 H, 3 Ph); ¹³C, δ 21.3 (q, OCO*C*H₃), 63.1 (t, C-6), 68.6, 72.9, 77.3 (3 d, C-3,4,5), 87.3 (s, *C*Ph₃), 99.2 (d, C-2), 127.4, 128.2, 129.0 (3 d, aromatic), 144.2 (s, aromatic), 146.5 (d, C-1), 171.9 (s, OCOCH₃).

Anal. Calc. for C₂₇H₂₆O₅: C, 75.33; H, 6.09. Found: C, 75.09; H, 6.18.

Compound **9** had $[\alpha]_D$ +51° (*c* 1.25, methanol). N.m.r. data: 1H , δ 1.88 (s, 3 H, OAc), 2.82 (bs, 1 H, HO-3), 3.17 (dd, 1 H, $J_{6.6}$ 10, $J_{5.6}$ 4 Hz, H-6), 3.48 (dd, 1 H, $J_{5.6}$ 3 Hz, H-6′), 4.08 (ddd, 1 H, $J_{4.5}$ 8 Hz, H-5), 4.13–4.35 (m, 1 H, H-3), 4.88 (dd, 1 H, $J_{1.2}$ 6, $J_{2.3}$ 1.8 Hz, H-2), 5.13 (dd, 1 H, $J_{3.4}$ 6 Hz, H-4), 6.55 (dd, 1 H, H-1); ${}^{13}C$, δ 20.9 (q, OCO*CH*₃), 62.4 (t, C-6), 67.3, 72.9, 75.6 (3 d, C-3,4,5), 87.2 (s, *CPh*₃), 102.7 (d, C-2), 127.4, 128.1, 129.1 (3 d, aromatic), 143.9 (s, aromatic), 144.9 (d, C-1), 171.1 (s, O*C*O*CH*₃).

Anal. Found: C, 75.22; H, 6.19.

Compound **10** had $[\alpha]_D$ +60° (*c* 1.2, methanol). N.m.r. data: 1 H, δ 1.80, 1.86 (2 s, 6 H, 2 OAc), 3.0–3.4 (m, 2 H, H-6,6′), 4.0–4.33 (m, 1 H, H-5), 4.71 (dd, 1 H, $J_{1,2}$ 6, $J_{2,3}$ 3.0 Hz, H-2), 5.0–5.38 (m, 2 H, H-3,4), 6.44 (d, 1 H, H-1), 7.0–7.7 (m, 15 H, 3 Ph); 13 C, δ 20.9, 21.1 (2 q, OCO*C*H₃), 62.1 (t, C-6), 67.8, 68.2, 76.0 (3 d, C-3,4,5), 87.2 (s, *C*Ph₃), 98.8 (d, C-2), 127.4, 128.1, 129.1 (3 d, aromatic), 144.1 (s, aromatic), 146.4 (d, C-1), 171.8, 172.1 (2 s, OCOCH₃).

Anal. Calc. for C₂₉H₂₈O₆: C, 73.71; H, 5.97. Found: C, 73.63; H, 5.99.

Compound **11** had $[\alpha]_D$ -88° (*c* 0.14, methanol). N.m.r. data: ${}^1\text{H}$, δ 2.75 (bs, HO-4), 3.33–3.66 (m, 2 H, H-6,6′), 3.88–4.33 (m, 2 H, H-4,5), 4.88 (dd, 1 H, $J_{1,2}$ 6, $J_{2,3}$ 2.5 Hz, H-2), 5.55 (ddd, 1 H, $J_{1,3}$ 1.8, $J_{3,4}$ 8 Hz, H-3), 6.55 (dd, 1 H, H-1), 7.11–7.77 (m, 15 H, 3 Ph); ${}^{13}\text{C}$, δ 63.0 (t, C-6), 68.6, 73.4, 77.9 (3 d, C-3,4,5), 87.5 (s, *C*Ph₃), 99.2 (d, C-2), 127.4, 128.2, 128.7, 129.1, 130.5, 133.4, 144.2 (aromatic), 146.7 (d, C-1), 168.1 (s, *OCOPh*).

Anal. Calc. for C₃₂H₂₈O₅; C, 78.03; H, 5.73. Found: C, 78.08; H, 5.81.

Compound **13** had $[\alpha]_D$ -11° (*c* 2.7, methanol). N.m.r. data: 1H , δ 3.31–3.57 (m, 2 H, H-6,6'), 4.26–4.55 (m, 1 H, H-5), 5.02 (dd, 1 H, $J_{1,2}$ 6, $J_{2,3}$ 3 Hz, H-2), 5.60 (ddd, 1 H, $J_{3,4}$ 6, $J_{1,3}$ 1.8 Hz, H-3), 5.73 (t, 1 H, $J_{4,5}$ 6 Hz, H-4), 6.62 (dd, 1 H, H-1), 7.05–8.35 (m, 25 H, 5 Ph); ${}^{13}C$, δ 62.1 (t, C-6), 68.3, 68.4, 76.1 (3 d, C-3,4,5), 87.1 (s, *C*Ph₃), 98.7 (d, C-2), 127.0, 127.3, 127.9, 128.9, 130.6, 133.1, 134.5, 143.9 (aromatic), 146.4 (d, C-1), 162.5, 168.2 (2 s, OCOPh).

Anal. Calc. for C₃₉H₃₂O₆: C, 78.50; H, 5.41. Found: C, 78.29; H, 5.48.

3-O-Acetyl- (16) and 3-O-benzoyl-L-rhamnal (19). — To a suspension of 15 (0.73 g, 2 mmol) in N,N-dimethylformamide (1.5 mL; or CH₂Cl₂, 3 mL) was added Et₃N (0.45 mL, 3.2 mmol) and Ac₂O (0.3 mL, 3 mmol) or BzCl (0.35 mL, 3 mmol) at room temperature, and the mixture was stirred for 2 h. The solvent was removed and the residue was subjected to column chromatography (light petroleum-ethyl acetate, 20:1) to give syrupy 16 (0.28 g, 81%) and crystalline 19 (0.37 g, 78%).

Compound 16 had $[\alpha]_D$ +21° (c 2.1, chloroform); lit.⁷ $[\alpha]_D^{25}$ +21° (chloro-

form). N.m.r. data: 1 H, δ 1.35 (d, 3 H, $J_{5,6}$ 6 Hz, H-6,6,6), 2.11 (s, 3 H, OAc), 3.22 (bs, 1 H, HO-4), 3.57 (dd, $J_{3,4}$ 7, $J_{4,5}$ 9 Hz, H-4), 3.86 (dq, 1 H, H-5), 4.66 (dd, 1 H, $J_{1,2}$ 6, $J_{2,3}$ 2 Hz, H-2), 5.22 (ddd, 1 H, $J_{1,3}$ 1.8 Hz, H-3), 6.44 (dd, 1 H, H-1).

Compound **19** had m.p. 63–64° (from dichloromethane–hexane), $[\alpha]_D$ +121° (c 1, chloroform), lit. 7 m.p. 64–65°, $[\alpha]_D^{25}$ +120° (chloroform). N.m.r. data: 1 H, δ 1.44 (d, 3 H, $J_{5,6}$ 6 Hz, H-6,6,6), 3.33–4.22 (m, 3 H, H-4,5 and HO-4), 4.80 (dd, 1 H, $J_{1,2}$ 7, $J_{2,3}$ 3 Hz, H-2), 5.44 (ddd, 1 H, $J_{1,3}$ 1.7, $J_{3,4}$ 8 Hz, H-3), 6.48 (dd, 1 H, H-1), 7.20–8.33 (m, 5 H, Ph).

4-*O*-Acetyl-L-rhamnal (17) had $[\alpha]_D$ -40° (*c* 1.5, chloroform); lit.⁷ $[\alpha]_D^{25}$ -41° (chloroform). N.m.r. data: ¹H, δ 1.28 (d, 3 H, $J_{5.6}$ 6 Hz, H-6,6,6), 1.33 (bs, 1 H, HO-3), 2.13 (s, 3 H, OAc), 3.75–4.07 (m, 1 H, H-5), 4.10–4.37 (m, 1 H, H-3), 4.75 (t, 1 H, $J_{3,4} = J_{4,5} = 9$ Hz, H-4), 4.79 (dd, 1 H, $J_{1,2}$ 6, $J_{2,3}$ 2 Hz, H-2), 6.37 (dd, 1 H, H-1).

4-*O*-Benzoyl-L-rhamnal (**20**) had m.p. 75–76° (from dichloromethane-hexane), $[\alpha]_{\rm D}$ –78° (*c* 1.1, chloroform); lit.⁷ m.p. 76°, $[\alpha]_{\rm D}^{25}$ –79° (chloroform). N.m.r. data: ¹H, δ 1.42 (d, 3 H, $J_{5,6}$ 6 Hz, H-6,6,6), 3.94 (dq, $J_{4,5}$ 8 Hz, H-5), 4.91 (dd, 1 H, $J_{1,2}$ 7, $J_{2,3}$ 3 Hz, H-2), 4.95 (m, 1 H, H-3), 5.48 (t, $J_{3,4}$ 8 Hz, H-4), 6.48 (dd, 1 H, H-1), 7.22–8.25 (m, 5 H, Ph).

3,4-Di-*O*-benzoyl-L-rhamnal (**21**) had $[\alpha]_{\rm D}$ +231° (*c* 1.3, chloroform); lit.⁷ $[\alpha]_{\rm D}^{25}$ +229° (chloroform). N.m.r. data: ¹H, δ 1.44 (d, 3 H, $J_{5,6}$ 6 Hz, H-6,6,6), 4.32 (dq, 1 H, $J_{4,5}$ 3 Hz, H-5), 4.94 (dd, 1 H, $J_{1,2}$ 7, $J_{2,3}$ 3 Hz, H-2), 5.41 (t, $J_{3,4}$ 8 Hz, H-4), 5.65 (ddd, 1 H, $J_{1,3}$ 1.7 Hz, H-3), 6.50 (dd, 1 H, H-1), 7.10–8.45 (m, 10 H, 2 Ph).

REFERENCES

- 1 I. D. BLACKBURNE, P. M. FREDRICKS, AND R. D. GUTHRIE, Aust. J. Chem., 29 (1976) 381-391.
- 2 J. S. Brimacombe, I. Da Aboul. and L. C. N. Tucker, Carbohydr. Res., 19 (1971) 276-280.
- 3 T. MAKI AND S. TEJIMA, Chem. Pharm. Bull., 15 (1967) 1367-1372.
- 4 K. K. OGILIVE, K. L. SADANA, E. A. THOMSON, M. A. QUILLIAM, AND J. B. WESTMORE, Tetrahedron Lett., (1974) 2861–2863.
- 5 B. Kraska, A. Klemer, and H. Hagedorn, Carbohydr. Res., 36 (1974) 398-403.
- 6 D. HORTON, W. PRIEBE, AND O. VARELA, Carbohydr. Res., 144 (1985) 325–330.
- 7 D. HORTON, W. PRIEBE, AND O. VARELA, Carbohydr. Res., 144 (1985) 317-324.
- 8 D. WAGNER, J. P. H. VERHEYDEN, AND J. G. MOFFATT, J. Org. Chem., 39 (1974) 24-30.
- 9 S. DAVID, A. THIEFRY, AND A. VEYRIERES, J. Chem. Soc., Perkin Trans. 1, (1981) 1796-1801.
- 10 Y. TSUDA, M. E. HAQUE, AND K. YOSHIMOTO, Chem. Pharm. Bull., 31 (1983) 1612-1616.
- 11 M. A. NASHED AND L. ANDERSON, Tetrahedron Lett., (1976) 3503-3506.
- 12 C. Auge, S. David, and A. Veyrieres, J. Chem. Soc., Chem. Commun., (1976) 375-376.
- 13 M. A. NASHED AND L. ANDERSON, Carbohydr. Res., 56 (1977) 325-336.
- 14 S. DAVID AND A. THIEFRY, J. Chem. Soc., Perkin Trans. 1, (1979) 1568–1573.
- 15 Y. TSUDA, M. HANAJIMA, AND K. YOSHIMOTO, Chem. Pharm. Bull., 31 (1983) 3778–3780.
- 16 Y. TSUDA, N. MATSUSHIRA, AND K. KANEMITSU, Chem. Pharm. Bull., 33 (1985) 4095-4097.
- 17 P. M. COLLINS (Ed.), Carbohydrates, Chapman and Hall, London, 1987, p. 432.